



**NAMIBIA UNIVERSITY  
OF SCIENCE AND TECHNOLOGY**

**FACULTY OF HEALTH, APPLIED SCIENCES AND NATURAL RESOURCES**

**DEPARTMENT OF NATURAL AND APPLIED SCIENCES**

<b>QUALIFICATION : BACHELOR OF SCIENCE</b>	
<b>QUALIFICATION CODE: 07BOSC</b>	<b>LEVEL: 7</b>
<b>COURSE CODE: SSP701S</b>	<b>COURSE NAME: SOLID STATE PHYSICS</b>
<b>SESSION: JULY 2022</b>	<b>PAPER: THEORY</b>
<b>DURATION: 3 HOURS</b>	<b>MARKS: 100</b>

<b>SUPPLEMENTARY/SECOND OPPORTUNITY EXAMINATION QUESTION PAPER</b>	
<b>EXAMINER(S)</b>	Prof Dipti R. Sahu
<b>MODERATOR:</b>	Dr Zivayi Chiguvare

<p style="text-align: center;"><b>INSTRUCTIONS</b></p> <ol style="list-style-type: none"><li>1. Answer all five questions.</li><li>2. Write clearly and neatly.</li><li>3. Number the answers clearly.</li></ol>
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**PERMISSIBLE MATERIALS**

Non-programmable Calculators

**THIS QUESTION PAPER CONSISTS OF 3 PAGES** (including this front page)

- Question 1** **[20]**
- 1.1 Distinguish the Lattice energy and cohesive energy. (4)
- 1.2 In a crystal whose primitives are 0.12nm, 0.18nm and 0.2 nm, a plane (231)' has an intercept of 0.12 nm on the x-axis. Find the intercept on the y and z axes. (6)
- 1.3 (a) Calculate the next neighbour's distance in a body centred cubic crystal (5)
- (b) In a body centred cubic lattice, find the ratio of the nearest neighbour's distance to the next neighbour's distance. (5)

- Question 2** **[20]**
- 2.1 How are dispersion bonds formed in a molecular solid? (4)
- 2.2 How do you identify Ionics, covalent and metallic bonds in solids (6)
- 2.3 Draw and explain energy vs interatomic distance curve. Compare it with force vs interatomic distance graphs (10)

- Question 3** **[20]**
- 3.1 What causes phonons in lattice vibration? Which crystals exhibit optical phonon modes? (4)
- 3.2 If the velocity of sound in a solid is of the order  $10^3$  m/s, compare the frequency of the sound wave with  $\lambda = 20 \text{ \AA}$  for (i) a monoatomic system and (ii) acoustic waves and optical waves in a diatomic system containing two identical atoms ( $M=m$ ) per unit cell of interatomic spacing  $2.2 \text{ \AA}$ . (6)
- 3.3 What is lattice specific heat? Drive heat capacity of solid crystalline substances at room temperature is  $3 R$ . (10)

- Question 4** **[20]**
- 4.1 Explain the terms: a) Drift velocity b) Relaxation time c) Mean free path d) Mean collision time for free electrons. (4)
- 4.2 The density and atomic weight of Cu are  $8900 \text{ kg.m}^{-3}$  and 63.5. The relaxation time of electrons in Cu at 300K is  $10^{-14}$ s. Calculate the electrical conductivity of copper. (6)
- 4.3 Derive Ohm's law based on Drude's Free Electron Theory? (10)

**Question 5****[20]**

- 5.1 On the basis of band structure how solids are classified? (4)
- 5.2 The Hall coefficient of certain silicon specimen was found to be  $-7.35 \times 10^{-5} \text{ m}^3 \text{ C}^{-1}$  from 100 to 400 K. Determine the nature of the semiconductor. If the conductivity was found to be  $200 \text{ } \Omega^{-1} \text{ m}^{-1}$ . Calculate the density and mobility of the charge carrier. (6)
- 5.3 What is an energy band? Why does the Fermi level in an intrinsic semiconductor lie in the middle of the energy gap? (10)
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**Given fundamental constants**Speed of light =  $3 \times 10^8 \text{ m/s}$ Planck constant =  $6.626 \times 10^{-34} \text{ Js}$ Mass of electron =  $9.1 \times 10^{-31} \text{ kg}$ Charge of electron =  $1.6 \times 10^{-19} \text{ C}$ Avogadro's number =  $6.022 \times 10^{23} / \text{mole}$ Boltzmann Constant =  $1.38 \times 10^{-23} \text{ JK}^{-1}$ 

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